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Facile hydrothermal synthesis of $\text{Ni}(\text{OH})_2$ -graphene foam composite for supercapacitor application

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Abstract content (Max 300 words) http://events.saip.org.za/getFile.py?target=_blank **Formatting** **Special chars**

Recently, one of the greatest scientific and engineering problems is the realization of a highly efficient energy and storage systems. Electrode materials are key to enhancing performance in a number of important energy storage technologies such as supercapacitors and batteries. Electrochemical capacitors also known as supercapacitors or ultracapacitors have become attractive and suitable in energy storage systems due to their extremely high power performance, moderate energy density and excellent cycle life. $\text{Ni}(\text{OH})_2$ has been identified as one of the most interesting transition metal hydroxide material owing to its easy synthesis, low cost and high theoretical capacitance. Thus, the design and synthesis of the nanoscale $\text{Ni}(\text{OH})_2$ based electrodes for high performance supercapacitors have attracted many attentions. In this work, we explore the synthesis of $\text{Ni}(\text{OH})_2$ -graphene foam (GF) composite via a facile hydrothermal reflux technique, and have investigated its potential use for supercapacitor application. The results obtained from the scanning electron microscopy showed that $\text{Ni}(\text{OH})_2$ spheres were uniformly distributed on the surface of the graphene foam. The specific capacitance of $\text{Ni}(\text{OH})_2$ -GF composite electrode was found to be 2420 F/g at a current density of 1 A/g with a coulombic efficiency of ~93% after 1000 charge/discharge cycles, demonstrating excellent cycle stability in 6.0 M KOH aqueous electrolyte. These results suggest that $\text{Ni}(\text{OH})_2$ -GF composite could be a potential electrode material for high performance electrochemical applications.

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yes

Level for award (Hons, MSc, PhD, N/A)?

PhD

Main supervisor (name and email) and his / her institution

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