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Red emission of Pr³⁺ enhancement by addition of Ba²⁺ and In³⁺ in CaTiO₃:Pr phosphor

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CaTiO₃ exhibits a Perovskite structures that has an ABO₃ stoichiometry where A and B represent metal cations with a combined charge of 6+ valency and O with 6- valency, and thus bring about charge neutrality to the compound. Upon introducing Pr³⁺ ions in the site of Ca²⁺ ions, charge imbalance occurs by introducing a 7+ valency, and this force the material to compensate for this effect by introducing negative Ca Ti vacancies, and positive O vacancies. These act by hampering the material luminescence. Thus substitution of Ti⁴⁺ by co-doping with Ba²⁺ and In³⁺ ions result into a 6+ valency and thus an electrically neutral compound. This reinforces the red emission of Pr³⁺ ions inside CaTiO₃. CaTiO₃: Pr³⁺ exhibiting red emission is synthesized via solid state reaction, and its luminescence is enhanced by co-doping with Ba²⁺ and In³⁺ that act as charge compensators. This optimal emission of CaTiO₃:Pr³⁺,Ba²⁺,In³⁺ is achieved by solid state reaction at 1300 oC for 4h, by direct firing, according to 1:1 molar ratio of TiO₂ Anatase and CaCO₃ compounds, doping with 0.3 mol

Level (Hons, MSc, PhD, other)?

MSc

Consider for a student award (Yes / No)?

Yes

**Would you like to
 submit a short paper
 for the Conference
 Proceedings (Yes / No)?**

Yes

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