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## Luminescence properties of blue-red emitting $\text{SrAl}_x\text{O}_y:1 \text{ percent Eu}^{2+}, x \text{ percent Cr}^{3+}$ phosphors prepared using sol-gel method

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### Abstract content <br> &nbsp; (Max 300 words)

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#### Abstract

$\text{SrAl}_x\text{O}_y:1\%\text{Eu}^{2+}, x\%\text{Cr}^{3+}$  powders were synthesized by sol-gel method at a low temperature below ( $\sim 80^\circ\text{C}$ ). Metal nitrates were used as the source of metal ions and citric acid as a chelating agent. Concentrations of  $\text{Cr}^{3+}$  ions were varied in the precursor during synthesis. The annealed samples were characterized by X-ray diffraction (XRD), scanning electron microscopy (SEM) and photoluminescence (PL) spectroscopy. The XRD revealed that the annealed samples consist of mixture of orthorhombic, unknown, cubic and hexagonal phases. Varying the  $\text{Cr}^{3+}$  concentrations do not affect the crystal structure of the phosphor. Morphology of the phosphor was influenced by varying the  $\text{Cr}^{3+}$  concentrations in the precursors. The PL results showed strong luminescence in the blue to red regions of the spectrum. The highest emission peak is at 702 nm, which is attributed to the typical  $4\text{T}_2(4\text{F}) - 4\text{A}_2(4\text{F})$  chromophore ( $\text{Cr}^{3+}$ ) ion transition. It was observed from the PL spectra that  $\text{Eu}^{2+}$  ions emissions were being suppressed relative to those of  $\text{Cr}^{3+}$  possibly due to energy transfer from  $\text{Eu}^{2+}$  ions to  $\text{Cr}^{3+}$  ions.

Keywords: Sol-gel,  $\text{SrAl}_x\text{O}_y:1\%\text{Eu}^{2+}, x\%\text{Cr}^{3+}$ , luminescent, energy transfer

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