







High-Pressure dependence of structure evolution and adsorption behavior in Nano-Layered Double Hydroxides (LDHs): impact for their uses in Cements.

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NextGenerationEU









VITALITY research project:

"Engineered sustainable concrete based on nanomaterials & recycled demolition waste"

(Principal Investigator: Prof. Paola Comodi)

DIPARTIMENTO DI FISICA E GEOLOGIA

Prof. Paola comodi Dr. Azzurra Zucchini Dr. Alessandro Di Michele Prof. Francesco Frondini DIPARTIMENTO DI INGEGNERIA

Prof. Emanuela Speranzini

DIPARTIMENTO DI SCIENZE FARMACEUTICHE

Prof. Aurelie Schoubben Prof. Riccardo Vivani DIPARTIMENTO DI CHIMICA, BIOLOGIA E BIOTECNOLOGIE

Prof. Paola Sassi

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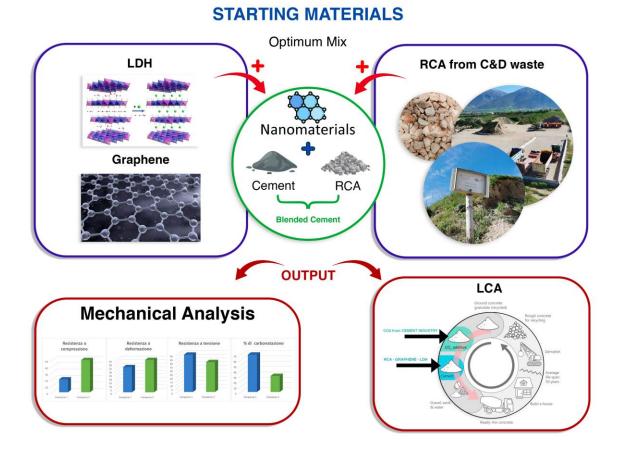




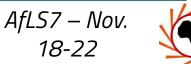




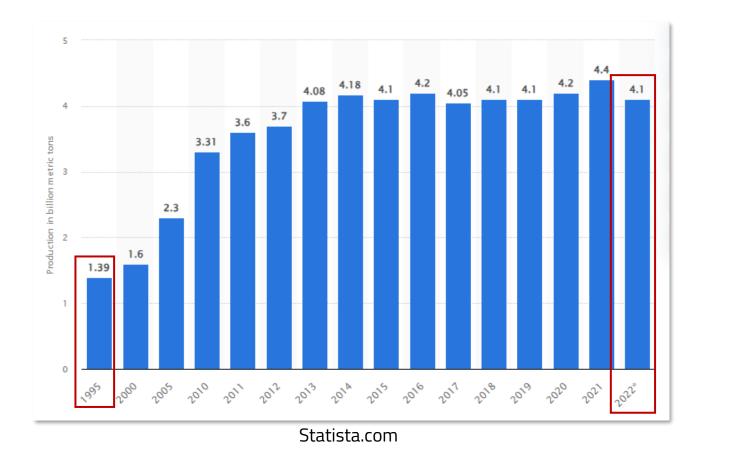
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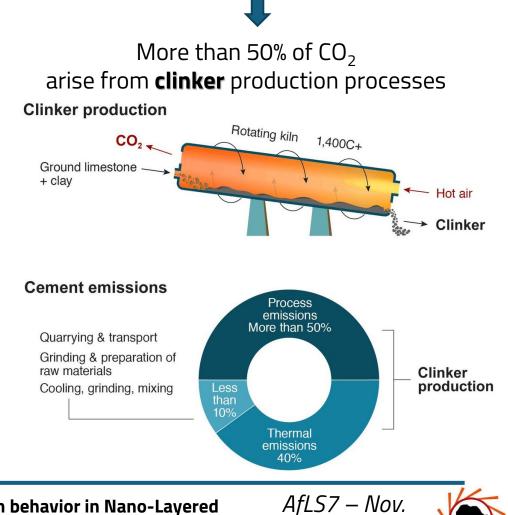
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In recent decade, demand and production of **cement** significantly **increased**



Cement production is responsible for approximately **7%** of global **CO₂** emissions



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Cement issue



 Corrosion of steel rebars caused by aggressive ions such as Cl⁻, CO₃⁻²and SO₄⁻ in reinforced concrete

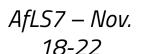


In alkaline environment, such as that of concrete, the steel bars a secured by a thin protective layer





In harsh environment, acids can react with various cement hydration products so that the pH value of concrete pore solution is significantly reduced





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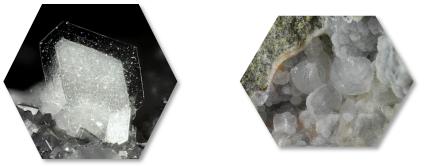
It is essential the development of new **strategies**

Incorporation of appropriate **nano-materials** into standard cement pastes

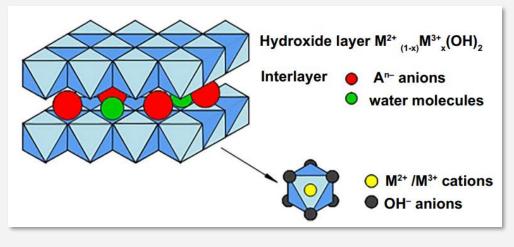
Improvement of **physical** and **chemical properties**

Layered double hydroxides (LDHs)

Also referred to as **anionic clays**, are emerging as excellent additives, thanks to their **structure** and interesting **properties**



LDHs consist of **positively charged** layers with intercalated **anions** in the inter-layer spaces.

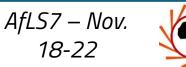


 $[M^{2+}_{1-x} M^{3+}_{x} (OH)_{2}]^{x+} (A^{n-})_{x/n} \cdot nH_{2}O$

M²⁺ = divalent cation M³⁺ = trivalent cation A = inter-layer anion

x = cations molar fraction

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LDH Characteristics and properties

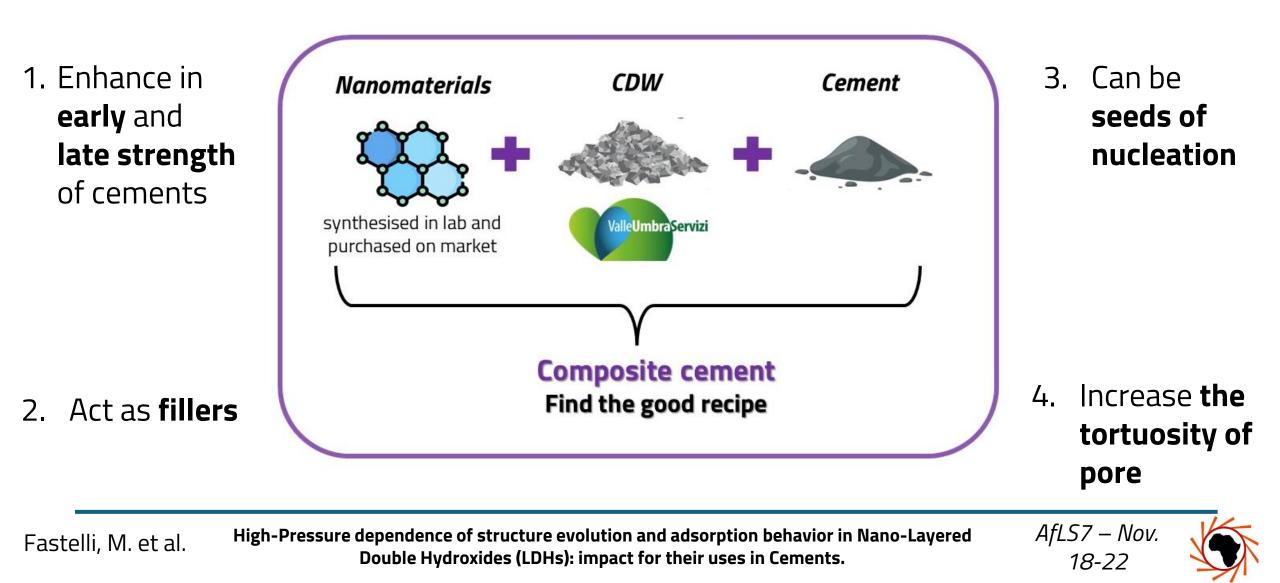
Environmental applications of LDHs Gas adsorption [adsorbents] $[M_{1-x}^{2+}M_x^{3+}(OH)_2]^{x+}$ c-parameter = 3 x *d* (3R symmetry **Recovery of CO**₂ [A_{x/n}]ⁿ⁻.mH₂O Ion exchange (including capture through Basal spacing (d) reconstruction) [adsorbent] LDHS Removal of ionic contaminants Interlamellar space and selective ion tuning Reduce porosity and great specific surface Н Anion, Aⁿ⁻ H₂O [nanomaterials] M^{2+}/M^{3+} а Improve mechanical properties Li et al., 2017 of concrete

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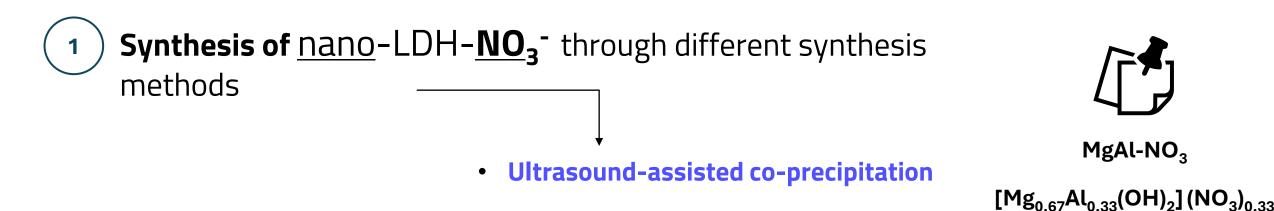
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LDH *inside cement paste*



Workflow



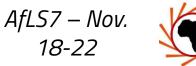
2

Characterization of the synthesized samples through

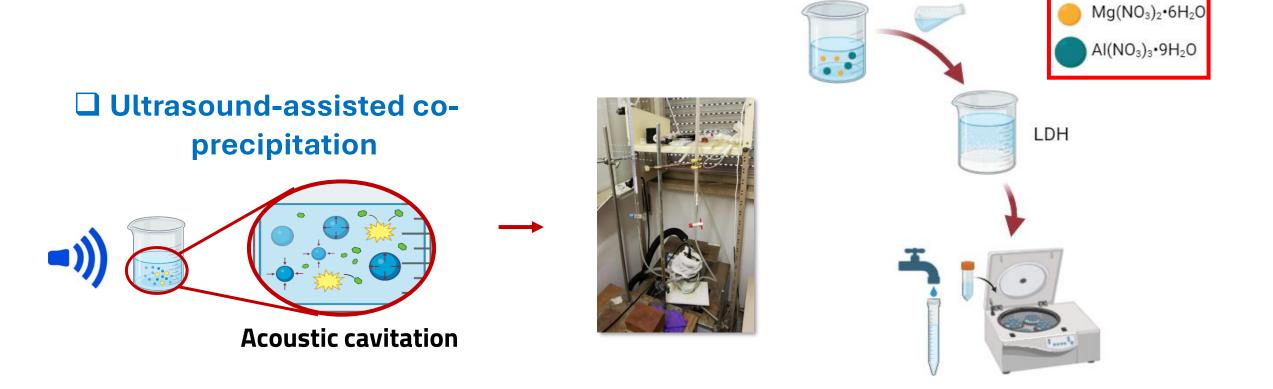
multiple analytical techniques

- **XRPD** (X-Ray Powder Diffraction)
- **FT-IR** (Infrared spectroscopy)
- **TGA** (Thermal Gravimetry Analysis)
 - **SEM** (Scanning Electron Microscopy)
 - **TEM** (Transmission Electron Microscopy)
 - **DLS** (Dynamic Light Scattering)

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Synthesis of <u>nano</u>-LDH-<u>NO₃</u>



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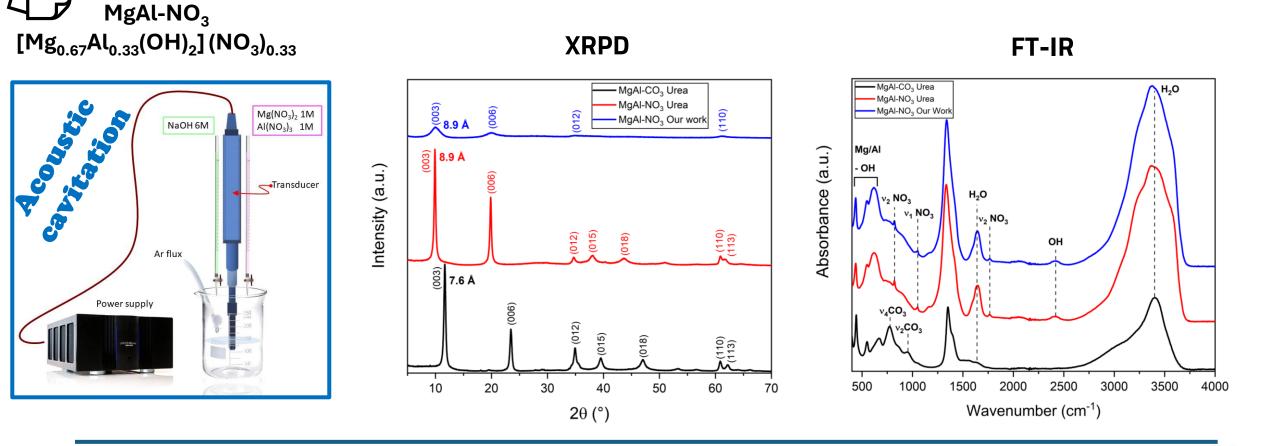
Na(OH)

Results: XRPD and FT-IR

2) Ultrasound assisted direct co-precipitation in supersaturation conditions

1 M solution of nitrate salts precipitated by 6M NaOH under argon atmosphere

followed by 24h ageing time at 80°C.



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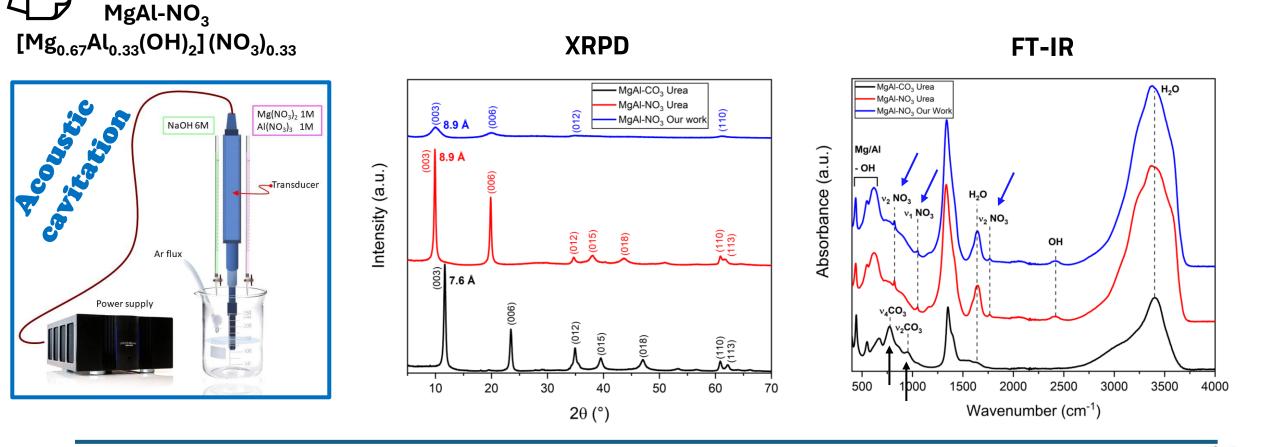
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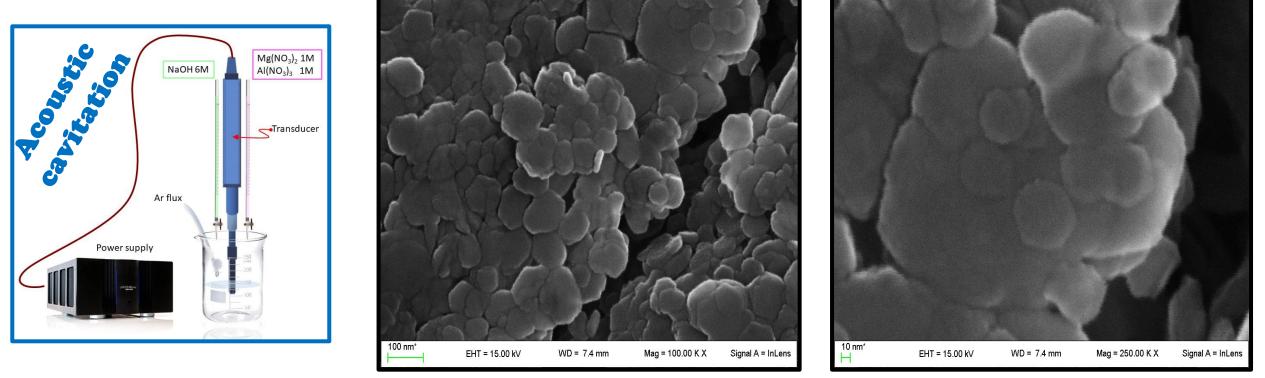
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MgAl-NO₃ [Mg_{0.67}Al_{0.33}(OH)₂] (NO₃)_{0.33}



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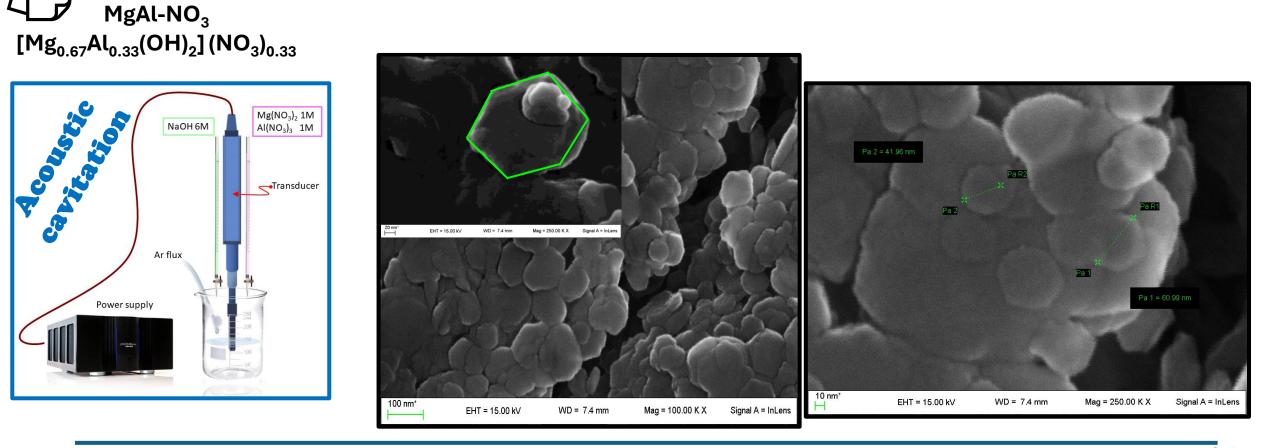
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Workflow



Preliminary study of the HIGH-PRESSURE powder diffraction behaviour of LDH-CO₃²⁻



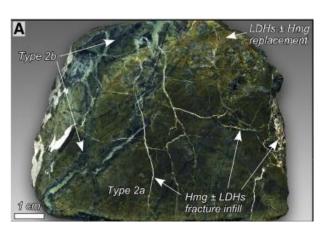
@ ESRF beamline ID 15b





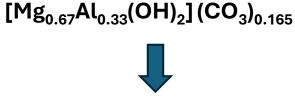
Science.org





Modified from Boschi et al., 2017

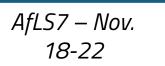




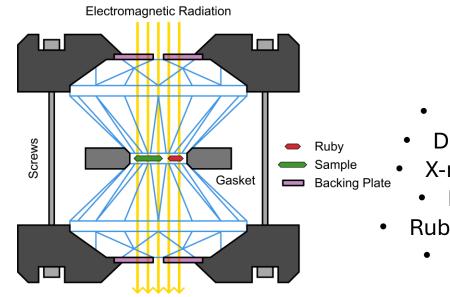
From Urea Synthesis method

Grenoble, France

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High Pressure (**HP**) X-ray powder diffraction using Diamond Anvil Cell (**DAC**)



schematic representation of a **DAC**

https://en.wikipedia.org/wiki/Diamond_anvil_cell#/media/File:Diamond_Anvil_Cell_____Cross_Section.svg

Experiment set-up:

- Beamline ID-15b @ESRF
- Detector EIGER2 X 9M CdTe
- X-ray wavelength 0.410371 Å
 - Ne transmitting medium
- Ruby and Au pressure calibrants
 - 300 µm coulet diamond

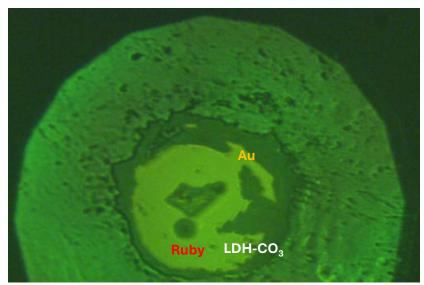
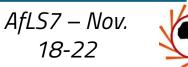
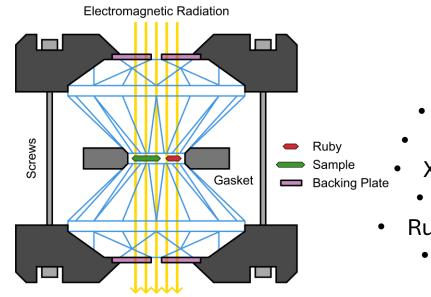


Photo of the inside of the **DAC**

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High Pressure (**HP**) X-ray powder diffraction using Diamond Anvil Cell (**DAC**)



schematic representation of a **DAC**

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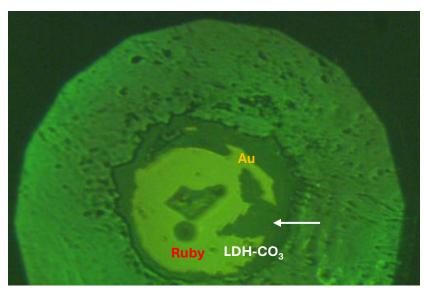
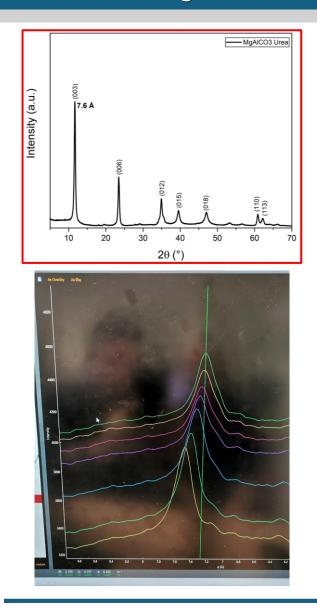


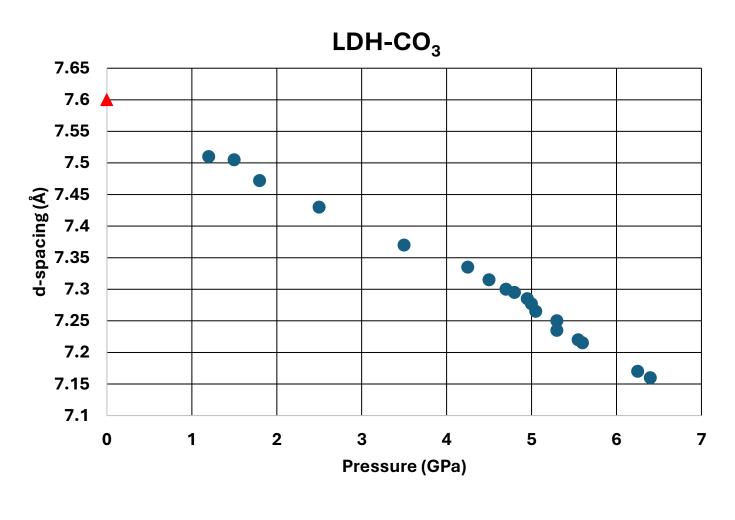
Photo of the inside of the **DAC**

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No amorphisation until 6.4 GPa

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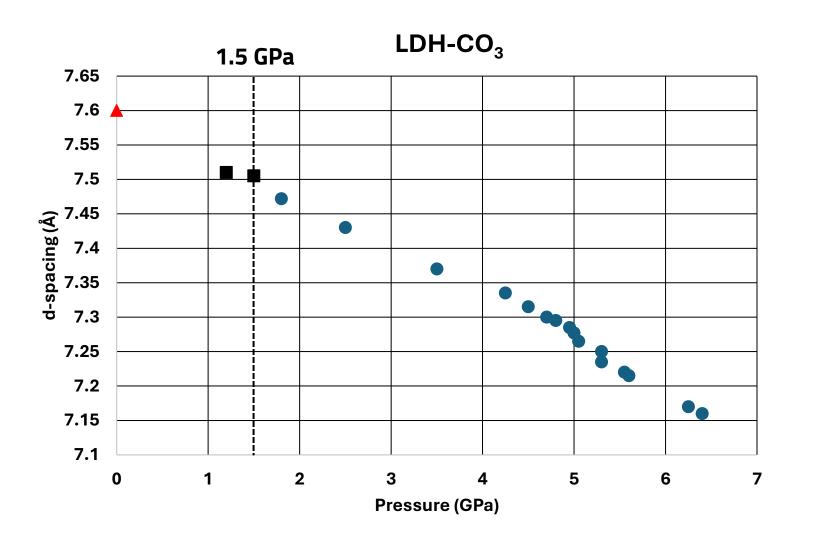
High-Pressure dependence of structure evolution and adsorption behavior in Nano-Layered Double Hydroxides (LDHs): impact for their uses in Cements. AfLS7 – Nov. 18-22 **1.5 Gpa: First transition** as
reported in Parthasarathy et al.,
2002 where were observed a
decrease of electrical resistivity



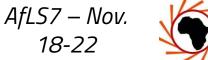
Conversion to **hydrotalcite** (3-layer rhombohedral polytype)

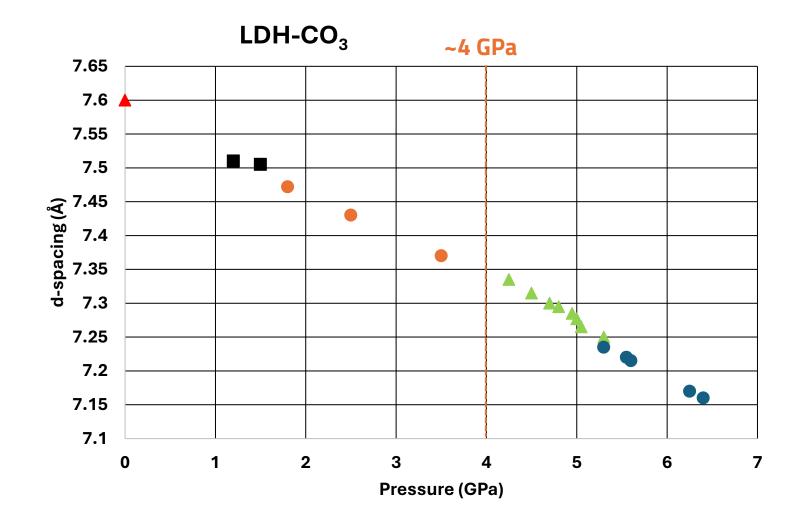
to

(Manasseite) polytype 2H (2-layer hexagonal polytype)



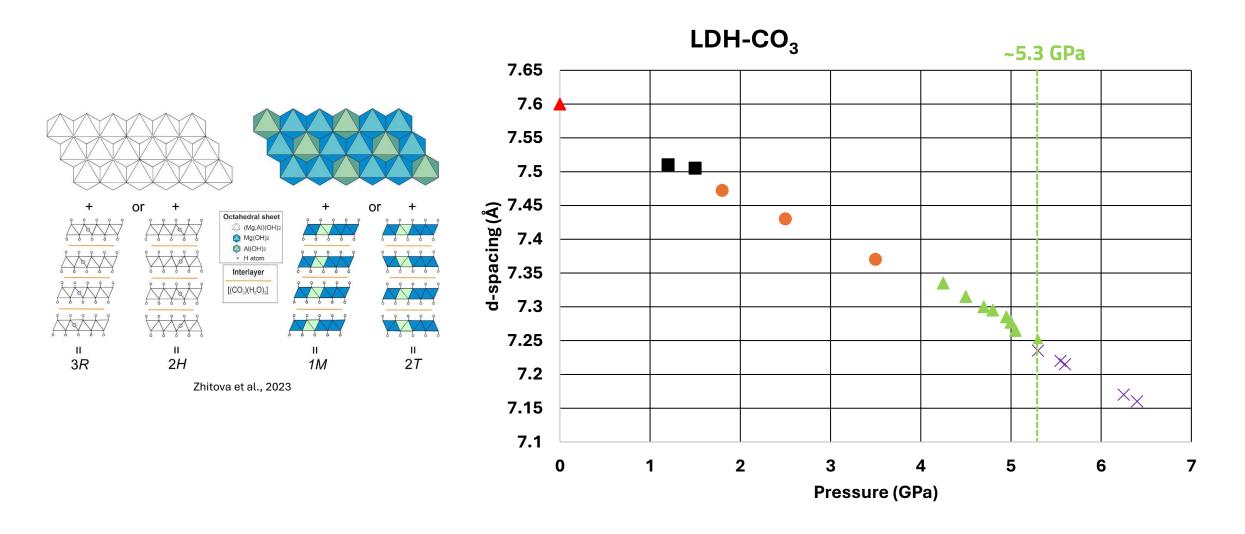
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No amosphisation of the sample were observed up to 6.4 GPa

A reduction of the peak intensity and broadening were observed

Given the **high quality of the data and resolution,** the observed **discontinuities could be significative** due to small ۲ error

> Discontinuity can be associated to a possible transition to polytype of hydrotalcite like materials or approach of the lamellas

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• Collection of a new dataset with a different set up

Using the DAC with diamond with a larger coulet and using a differentiated pressure medium that can be intercalated within the structure.

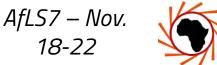
• Studying the effect of pressure not only by diffraction technique but also by *in situ* **Raman spectroscopy**

Raman spectroscopy could provide important information on possible phase transitions and intercalation of the pressure medium.

• Studying the contribution of pressure on the **memory effect** of LDH

Indeed, temperature does not affect the memory effect but the effect of pressure on this property is not known

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Thank you for your attention!

...and thanks to the whole research team





Zucchini A.

Di Michele A.



Mortaro F.

Schoubben A.





The European Synchrotron

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