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THE CATALYTIC ACTIVITY OF MO_2 (M= Ti, V, Mn) ON (110) SURFACE OF Na-AIR BATTERY

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Metal-air batteries have a far higher theoretical energy density than lithium-ion batteries and are frequently proposed as a possibility for next-generation electrochemical energy storage for applications like electric vehicles and grid energy storage. The Na-air battery is an energy storage system with moderate catalytic activity in the Oxygen Reduction Reaction (ORR) and Oxygen Evolution Reaction (OER). The current study uses the Density functional theory (DFT) to develop a deeper understanding of the catalytic activity of rutile types of MO_2 for the ORR and OER in Na-air batteries. As oxygen is adsorbed on Na/ MO_2 (110) surfaces, the electronic properties of several configurations (dissociated, superoxide, peroxy on Na-M, and dissociated) are examined to determine the catalytic influence towards the generation of the discharge product in Na- O_2 batteries. Using the density of states analysis, the MnO_2 system was found to be more stable than TiO_2 and VO_2 systems. These discoveries are significant for the nucleation and development of Na_2O_2 and NaO_2 , as well as for understanding the reactivity of Na/ MO_2 (110) surfaces.

Apply to be considered for a student ; award (Yes / No)?

YES

Level for award;(Hons, MSc, PhD, N/A)?

PhD

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