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## Effect of Gd<sup>3+</sup> moles on the structure, morphology and luminescence properties of BaAl<sub>2</sub>O<sub>4</sub>:x% Gd<sup>3+</sup> (0 ≤ x ≤ 1) nanomaterial prepared using a sol-gel method.

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In this study BaAl<sub>2</sub>O<sub>4</sub>:x%Gd<sup>3+</sup> (0 ≤ x ≤ 1) nanomaterial were prepared using sol-gel method. X-ray diffraction (XRD) data revealed that all samples consisted of a single phase of cubic BaAl<sub>2</sub>O<sub>4</sub> structure with no impurities. Fourier Transformation Infrared spectroscopy (FTIR) revealed four absorption bands at 843, 1016, 1416, and 3434 cm<sup>-1</sup>. The presence of Ba, Al, O and Gd were confirmed by the energy dispersive X-ray spectroscopy (EDS). Scanning electron microscope (SEM) revealed that the morphology of the prepared samples highly depends on the Gd<sup>3+</sup> concentration. Transmission electron microscopy (TEM) results revealed the tubular rods and nano-nature of the crystallite sizes. Photoluminescence (PL) spectroscopy results revealed three emission peaks located at 414 nm, 436 nm, and 748 nm. All the emissions are from the host, all these emissions are attributed to arise from the intrinsic defects within the host material such as oxygen. The results revealed the Gd<sup>3+</sup> optimum intensity at 0.6% Gd<sup>3+</sup>. This suggests that increasing the concentration of Gd<sup>3+</sup> in the host influences the luminescence of the nano-powders. The International Commission on Illumination (CIE) colour showed that the violet emission colour of the prepared samples depends on the excitation wavelength and Gd<sup>3+</sup> concentration.

### Apply to be considered for a student ; award (Yes / No)?

No

### Level for award;(Hons, MSc, PhD, N/A)?

N/A

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