

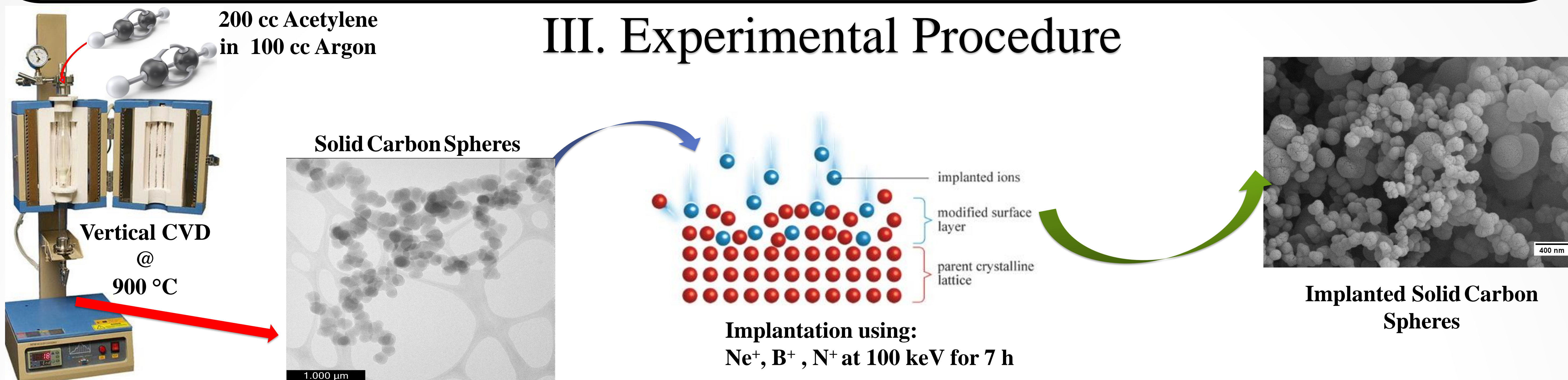
I. Introduction

Carbon-based nanostructures have been the forefront materials of interest for a vast array of applications since the discovery of fullerenes in the mid-1980's. Among the carbonaceous material family, solid carbon nanospheres (SCSs) have gained interest in the renewable energy technology arena due to their spherical morphological framework which provides accessible surface for charge storage. Moreover, the surface properties can be modulated by introduction of the heteroatoms. Practically, modulation to desired depths via the ion-implantation technique is the preferred route as the core of the SCSs nanostructures is not compromised, whilst the tuned surface is available for either energy conversion and/or storage applications.

II. Research Objectives

- Employ the vertical CVD technique for synthesis of SCSs.
- Post ion implantation using Ne^+ , B^+ , and N^+ ions.
- Investigate the influence of heteroatom ion on the physicochemical properties of SCSs.

III. Experimental Procedure

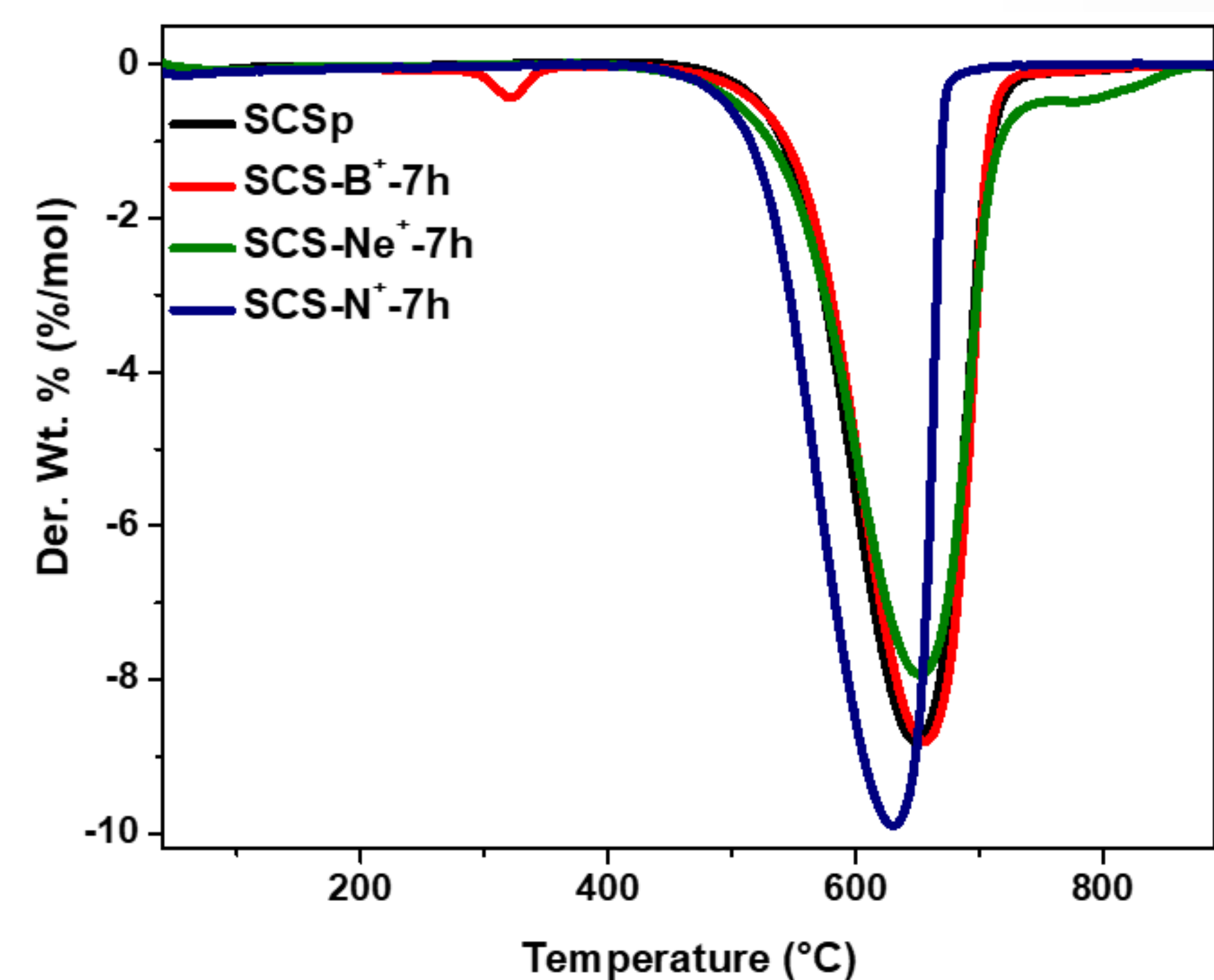
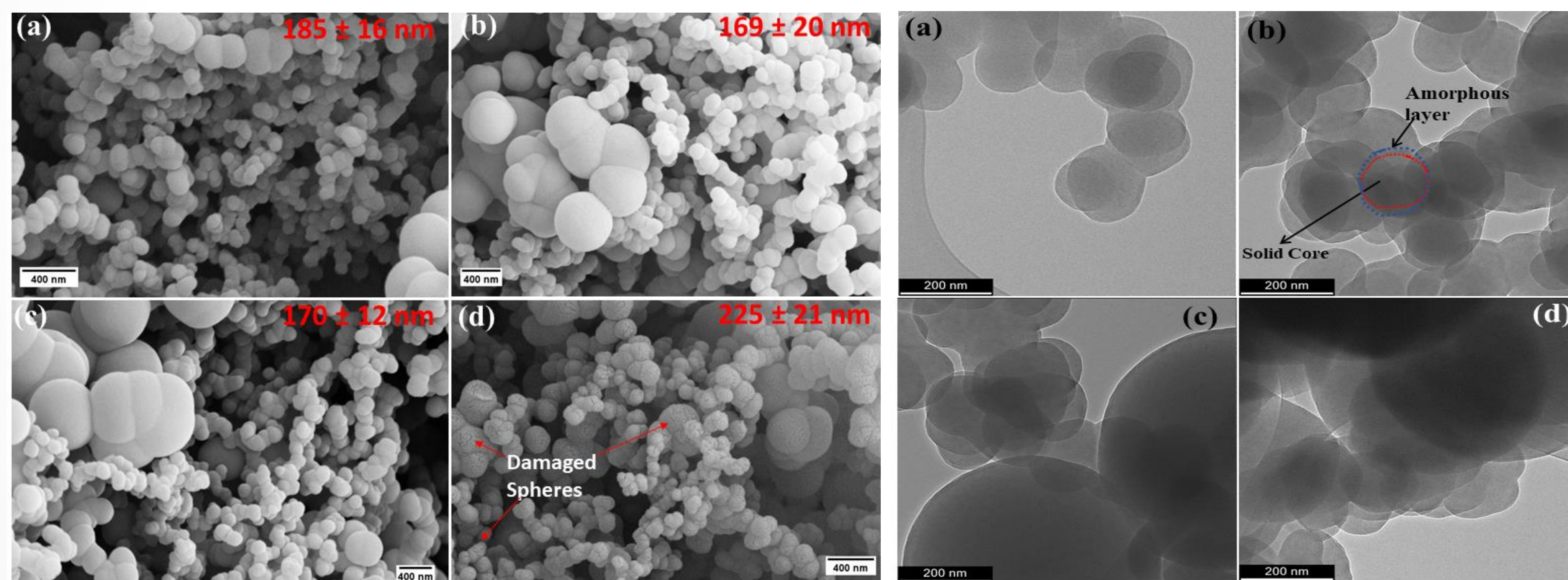


IV. Results and Discussions

(i) SEM micrographs

(ii) TEM micrographs

(iii) Thermal stability



(i) SEM and (ii) TEM images of (a) Pristine and doped (b) Ne^+ (c) B^+ , and (d) N^+ SCSs. Depiction of agglomerated spheres as well as damaged spheres for N-implanted samples. Presence of varying thickness of amorphous layer around SCSs depending on heteroatom ion (**Table 1**). Thicker amorphous layer suggestive of depth of implantation.

Pristine SCSs are more thermally stable due to solid core. Narrow bandwidth for ion-implanted samples signifies generation of defects, thus enabling easy decomposition and compromised thermal stability.

Table 1: Textual and morphological parameters of the samples

Sample	BET surface area (m^2/g)	Pore size (nm)	Amorphous layer (nm)	Decomp. Temp ($^{\circ}\text{C}$)
SCS	13.3	25.1	2.3 ± 0.5	650
Ne^+ - SCSs	13.6	51.9	7.8 ± 1.8	652
B^+ - SCSs	12.5	22.4	16.9 ± 2.2	321, 656
N^+ - SCSs	10.1	21.6	13.6 ± 1.8	630

V. Conclusions

- Successful growth of SCSs using vertical CVD technique and implantation with Ne^+ , B^+ , and N^+ ions.
- Physicochemical properties of ion-implanted samples was dependent on the type of heteroatom ion..
- N^+ -7h implantation led to more defective and less thermally stable samples.
- Affinity of B^+ and N^+ ion to carbon lattice led to increased outer shell distortion; suggestive of a deeper ion implantation.
- ❖ **Demonstration of significance of choice of heteroatom ion for tuneable physicochemical properties of SCSs.**



Acknowledgements

