

Contribution ID: 163

Type: Oral Presentations

Heterotrinuclear complex of (Ni-K-Ni) and Co-crystal structure of a dinuclear (Zn-Y) and a trinuclear (Zn-Y-Zn)

Wednesday, 30 January 2019 16:30 (15 minutes)

The very dynamic field of research for the design and synthesis of heteropolynuclear complexes of d and f-block elements has made it possible to generate original structures with interesting properties in various fields such as magnetism, optics, catalysis and luminescence. The present investigation describes the synthesis and structural study of two complexes. The complex 1 is a Heterotrinuclear complexes synthesis which was used to generate three dimensional complexes formulated as {Ni(L)}(K){Ni(L)}. The compound crystallizes in the Monoclinic system Space group P2 1 /c with the following unit cell dimensions a =11.5891(13) Å b = 17.307(2) Å c = 19.781(2) Å and b= 92.368(10)°, R1= 0.1172. In the compound the ligand H 2 L is hexadentate. In the unit the coordination environment of the Ni metal can be described as distorted square, the alkaline metal K is octacoordinated. The complex 2 is a co-crystal which was a three dimensional supramolecular complexs. formulated as [Y{Zn(L)(SCN)}(SCN) 2].[Y{Zn(L)(SCN)} 2 (DMF) 2].(NO 3). It crystallizes in the triclinic space group P-1 with unit cell parameters: a = 14.8987(7) Å, b = 15.6725(8) Å, c = 19.2339(10) Å, α = 94.610(4)°, β = 103.857(4)°, γ = 101.473(4)°, R1 = 0.063. Heterodinuclear unit [Y{Zn(L)(SCN)}(SCN) 2] is co-crystallized with a heterotrinuclear unit [Y{Zn(L)(SCN)} 2 (DMF) 2].(NO 3). In the dinuclear moiety, the ligand is hexadentate and in the trinuclear unit, it is pentadentate with one of the oxygen methoxy group remaining uncoordinated. The coordination environment Zn metal is square pyramidal. In the dinuclear unit the Y(III) is hexacoordinated while it is octacoordinated in the trinuclear unit. The environment of the Y(III) can be described as a distorted octahedral geometry in the dinuclear and as a distorted square antiprism in the trinuclear units respectively.

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Session Classification: PCCr2

Track Classification: PCCr2