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Contribution ID: 305

Type: **Poster Presentation**

Computer Simulation Study of Synthetic Nickel-Rich Pentlandite (Fe₄Ni₅S₈) Surfaces

Tuesday, 10 July 2012 17:30 (2 hours)

Abstract content (Max 300 words)

Density Functional Theory (DFT) with the plane-wave (PW) pseudo-potential method within the VASP code is employed to investigate the interaction of oxygen and water molecules on the synthetic nickel-rich pentlandite {100} and {110} surfaces. The oxygen-metal interaction show preferential iron oxidation than nickel on both surface models. However, a bridging of the oxygen molecule between the Fe-Ni metals on the {100} surface prefers the horizontal orientation with strong adsorption energy of -5.60 eV, while the un-oxidized nickel atoms dissociates into the sub-surface. We also observe a peroxo isomer anion (Fe (O₂)⁻) species on the {100} surface with one oxygen forming a bridging bond with the nearest nickel atom. Moreover, the iron preferential oxidation is clearly seen on the {110} surface with the vertical orientation, where it moves towards iron forming a bridge. The oxidation of the nickel atom showed the presence of superoxo isomer species on the {110} surface. The hydration of nickel on {100} surface gives the most stable surface compared to {110} with a strong adsorption energy of -2.96 eV (exothermic) and while endothermic process is observed for the {110} surface (0.14 eV). However, hydration of {110} surface, in particular on the iron atom shows weak adsorption energy -0.07 eV (exothermic).

Apply to be considered for a student award (Yes / No)?

yes

Level for award (Hons, MSc, PhD)?

MSc

Main supervisor (name and email) and his / her institution

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Would you like to submit a short paper for the Conference Proceedings (Yes / No)?

No

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Session Classification: Poster Session

Track Classification: Track A - Division for Condensed Matter Physics and Materials