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The thermal quenching process of the $\text{La}_2\text{O}_3\text{:Eu(III)}$ phosphor material

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This study is concentrating on the measurement of the emission of commercially available lanthanum oxy-sulphide doped with europium(III) ($\text{La}_2\text{O}_3\text{:Eu(III)}$) phosphor material at various temperatures. For the thermal quenching process, the average activation energies for the emission from the ${}^5\text{D}_2$, ${}^5\text{D}_1$ and ${}^5\text{D}_0$ excited states were determined as 0.49 eV, 0.55 eV and 0.77 eV, respectively and the average pre-exponential constant was determined as $9.5 \times 10^7 \text{ s}^{-1}$. The optical band gap of $\text{La}_2\text{O}_3\text{:Eu(III)}$ was determined as 2.75 eV. It was also established that the sulphur(II) to europium(III) (Eu(III)) charge transfer band absorbs ultraviolet light and transfers the excited electrons to the excited states of the Eu(III) ions from where emission can take place. The lifetime of the luminescence results indicated that the higher excited states have a double exponential lifetime that results from the emission from both the conventional Eu(III) ions and Eu(III) ions that are in the vicinity of a defect or impurity group. It was determined that in the case of the $\text{La}_2\text{O}_3\text{:Eu(III)}$ phosphor material, the presence of defect or impurity groups is due to the hydroxide groups that forms when the material was exposed to water vapour in the atmosphere at room temperature. The average emission decay constants of the ${}^5\text{D}_2$, ${}^5\text{D}_1$ and ${}^5\text{D}_0$ excited states were determined as 10 ns, 80 ns and 340 ns respectively. It was also revealed that $\text{La}_2\text{O}_3\text{:Eu(III)}$ can be utilised as a temperature sensor by using the fluorescence intensity ratio of the emission from the ${}^5\text{D}_1$ and ${}^5\text{D}_0$ excited states. This worked well for the temperature range from 80 °C to 180 °C.

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Hendrik C. Swart, Department of Physics, University of the Free State, Bloemfontein, South Africa, SwartHC@ufs.ac.za

Primary authors: Prof. SWART, Hendrik (University of the Free State); Prof. TERBLANS, JJ (Koos) (UFS); Mr ERASMUS, Lucas (University of the Free State); Dr CRACIUN, Valentin (National Institute for Laser, Plasma and Radiation Physics)

Presenter: Mr ERASMUS, Lucas (University of the Free State)

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