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## Analyses of the Pore Structure Formation of Activated Carbon from Compaction of Plant Biomass Waste for Electrochemical Device Electrodes

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The formation mechanism of the porous framework in nanostructured carbon materials is important in a wide variety of applications such as in supercapacitors, gas storage, adsorbents and catalyst supports etc. The accessibility to the pore sites by electrolyte ions and gases are highly determined by the precise synthesis techniques adopted for these materials. As such, biomass waste materials are a good choice for synthesis as they are available in abundance and cheap, while containing high carbon content and giving high specific surface area for electrochemical supercapacitor applications.

In this study, activated carbon (AC) was synthesized from renewable plant biomass waste using a chemical vapour deposition (CVD) technique via a pre-hydrothermal conversion step and compaction along with the fine-tuning of key growth parameters, such as activating agent and carrier gas. The textural, structural and morphological features were investigated by the Brunauer-Emmett-Teller (BET) technique, X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), and Scanning/Transmission electron microscopy (SEM/TEM) characterization. The material tested in a three-electrode configuration exhibited electric double-layer capacitor (EDLC) behaviour in different electrolytes and working comfortably in KNO<sub>3</sub> electrolyte in both negative and positive operating window of 0.80 V. The results from this study provide the pathway into designing hierarchically porous materials from cheap and sustainable sources suitable in high power energy storage devices.

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